

PII: S0960-894X(96)00612-9

## ANTIBODY-MEDIATED REGIO- AND ENANTIOSELECTIVE RESOLUTION OF A GLYCEROL DERIVATIVE

## Kiyoshi Ikeda and Kazuo Achiwa\*

School of Pharmaceutical Sciences, University of Shizuoka, Yada 52-1, Shizuoka 422, Japan

**Abstract:** The antibody-mediated regio- and enantioselective hydrolysis of a glycerol derivative is reported. (R)-2-Acetoxy-1-(3-nitrobenzyloxy)glycerol (80% ee) was obtained from the (R,S)-2,3-diacetoxy glycerol derivative in 36% yield at the antibody-catalyzed kinetic resolution step. © 1997, Elsevier Science Ltd. All rights reserved.

Optically active glycerol derivatives are useful as starting materials for the syntheses of several types of chiral medicines, including (S)-propranolol and related compounds, <sup>1</sup> PAF antagonists and their derivatives, <sup>2</sup> 4-amino-3-hydroxybutyric acid (GABOB)<sup>3</sup> as its simple derivatives, lipopeptides<sup>4</sup> and many biologically active compounds.<sup>5</sup> So far many synthetic methods for chiral glycerol derivatives have been established. Although the asymmetric syntheses of glycerol derivatives by enzymatic reactions have been documented, <sup>6</sup> kinetic resolution by an enzyme catalyst is still a useful method for the synthesis of optically active glycerol derivatives.

High levels of regio- and stereoselectivity are one of the greatest advantages of using an antibody catalyst.<sup>7</sup> Chemoselective hydrolysis of glycerol derivatives having several nearly equivalent reaction coordinates is particularly difficult to control. In this paper, we describe the antibody-mediated regio- and enantioselective hydrolysis of the (R,S)-2,3-diacetoxy glycerol derivative.

Synthesis of Hapten. We designed haptenic phosphonate transition state analog 12 to generate catalytic antibodies that enantioselectively hydrolyze (R,S)-1 to the alcohol (R)-2. Enantiomerically pure hapten 12 that possesses a phosphonate group mimicking that the tetrahedral intermediate for ester hydrolysis of a glycerol derivative was prepared in nine steps from (S)-2,3-O-isopropylidene glycerol 6, as outlined in Scheme 1. The hydroxyl group of 6 was protected as the p-nitrobenzyl group by treatment with p-nitrobenzyl bromide, Ag<sub>2</sub>O and t-butylammonium iodide (TBAI) in 49% yield, and then the isopropylidene group was hydrolyzed by 60% acetic acid to give 7 in 80% yield. The primary hydroxyl group of 7 was selectively protected by t-butyldimethylsilyl chloride (TBDMSCI) and imidazole in 81% yield, and then the protection of the remaining hydroxyl group was carried out with dihydropyran (DHP) and PPTS to afford 8 in 92% yield. Treatment of 8 with t-butylammonium fluoride (TBAF) gave monoalcohol 9 in 72% yield. Phosphorylation of the hydroxyl group of 9 by 5, having the five-carbon spacer as the linker in hapten synthesis, was successfully accomplished in the presence of NEt3 and DMAP to afford phosphonate 10 in 65% yield. Removal of the p-methoxybenzyl group and THP groups of 10 were simultaneously carried out with trifluoroacetic acid to give 11 in 79% yield. After acetylation of 11 with Ac<sub>2</sub>O and pyridine in 80% yield, treatment with trimethylsilyl bromide gave hapten 12 in 95% yield, FAB-MS (3-NBA matrix): m/z (M+H+) 448.

Conditions: i) 1) p-MeOC<sub>6</sub>H<sub>4</sub>CH<sub>2</sub>OH, DCC, DMAP, CH<sub>2</sub>Cl<sub>2</sub>, 0°C then rt, 15 h (yield 86%), 2) (MeO)<sub>3</sub>P, p-cymene, 170°C, 24 h (yield 74%); ii) (COCl)<sub>2</sub>, CH<sub>2</sub>Cl<sub>2</sub>, rt, 18 h (quant.).

Conditions: i) 1) p-NO<sub>2</sub>C<sub>6</sub>H<sub>4</sub>CH<sub>2</sub>Br, Ag<sub>2</sub>O, TBAI, MS4A, CH<sub>2</sub>Cl<sub>2</sub>,  $\pi$ , 15 h (yield 49%); 2) 60% AcOH, 80-90°C, 2 h (yield 80%); ii) 1) TBDMSCI, imidazole, CH<sub>2</sub>Cl<sub>2</sub>,  $\pi$ , 15 h (yield 81%); 2) DHP, PPTS, CH<sub>2</sub>Cl<sub>2</sub>,  $\pi$ , 15 h (yield 92%); iii) TBAF, THF,  $\pi$ , 1 h (yield 72%); iv) 5, NEt<sub>3</sub>, DMAP, 0°C then  $\pi$ , 15 h (Yield 95%); v) TFA-CH<sub>2</sub>Cl<sub>2</sub> (1:10),  $\pi$ , 2 h (yield 79%); vi) 1) Ac<sub>2</sub>O-pyridine (1:2),  $\pi$ , 24 h (yield 80%), 2) TMSBr, CH<sub>3</sub>CN, 40-50°C, 45 h (yield 95%); vii) 1) KLH or BSA,EDC in DMF-H<sub>2</sub>O, 2) dialysis, NaCl buffer, pH7.4.

Antibody Production. Hapten 12 was coupled to the carrier proteins keyhole limpet hemocyanin (KLH) and bovine serum albumin (BSA) using a water-soluble carbodiimide (EDC) to provide the corresponding protein conjugates. The precipitates were purified by chromatography on Sephadex G-25. The KLH conjugate was used as an antigen, and the BSA conjugate was used in ELISA experiments for measuring serum titer and hapten affinity. Balb/c mice were immunized with the KLH conjugate of 12, and hybridomas were prepared from the immunized spleenocytes using standard hybridoma protocols.<sup>8</sup> We obtained five stable hybridoma cell lines that exhibited binding specificity for BSA-12. Samples of monoclonal antibodies were prepared by *in vivo* ascites production and purified from ascites fluid to homogenecity by ammonium sulfate precipitation followed by protein G affinity chromatography. The antibodies were dialyzed in PBS at pH 7.4. The homogenecity of each antibody was determined by sodium dodecyl sulfate-polyacrylamide gel electrophoresis (SDS-PAGE) with Coomassie blue staining.

Catalytic Assay and Kinetics. Antibodies that bound to BSA-12 were screened for the ability to catalyze the hydrolysis of (R,S)-1. The reaction was performed using 1.3  $\mu$ M of antibody  $^9$  and 100-750  $\mu$ M of (R,S)-1 in 10% DMSO/0.2 M Tris (pH 8.0) at 30°C by monitoring production of (R)-210 by reverse-phase high-pressure liquid chromatography (HPLC) and N-ethylbenzamide as the internal reference  $^{11}$  As a result, three of five antibodies were found to accelerate, over background hydrolysis,  $^{12}$  the hydrolytic degradation

of (*R,S*)-1 to the alcohol (*R*)-2, and the most effective antibody 2D10 was characterized in further detail. Antibody 2D10 displayed saturation kinetics described by the Michelis-Menten equation in the hydrolysis of (*R,S*)-1. The kinetic parameters of 2D10 from the Lineweaver-Burk plot were afforded values of  $K_m$ =1.3 mM,  $V_{max}$ =2.6 mM min<sup>-1</sup> and  $k_{cat}$ =2.0 min<sup>-1</sup>, respectively. Furthermore, hydrolysis of (*R,S*)-1 by 2D10 is competitively inhibited with the addition of chiral phosphonate 12 ( $K_i$ = 2.8  $\mu$ M).

Kinetic Resolution. When the reaction was performed using antibody 2D10 (1.3  $\mu$ M) and (*R,S*)-1 (1 mM) in 5% DMSO/0.2 M Tris (pH 8.0) at 4°C, 80% ee of (*R*)-2 (36% hydrolysis conversion) was obtained. The enantiomeric excess of diester (*R*)-1 and monoester (*R*)-2 was measured by HPLC using a chiral column.<sup>13</sup>

OAC 
$$IgG 2D10$$
  $IgG 2D10$   $IgG 2$ 

 $K_{m} \ 1.3 \ mM, \ V_{max} \ 2.6 \ \mu M \ min^{-1}, \ k_{cat} \ 2.0 \ min^{-1}, \ K_{i} \ 2.8 \ \mu M$ 

 $R = p - NO_2C_6H_4CH_2$ 

**Conclusion.** We demonstrated the generation of antibodies that regio- and enantioselectively hydrolyze a glycerol derivative to afford the optically active glycerol derivative. This procedure represents an efficient method for the optical resolution of glycerol derivatives.

## References and Notes

- Tsuda, Y.; Yoshimoto, K.; Nishikawa, T. Chem. Pharm. Bull. 1981, 29, 3593; Iriuchijima, S.; Kojima, N. Agric. Biol. Chem. 1982, 46, 1153; Iriuchijima, S.; Keiyu, A.; Kojima, N. ibid. 1982, 46, 1593; Jung, M.E.; Shaw, T. J. J. Am. Chem. Soc. 1980, 102, 6304; Klunder, J. M.; Ko, S. Y.; Sharpless, K. B. J. Org. Chem. 1986, 51, 3710.
- 2. Shiraiwa, M.; Fujita, K.; Yoshiwara, H.; Kobayashi, S.; Ohno, M. J. Org. Syn. Chem. Japan 1987, 45, 369 and references cited therein.
- 3. Takano, S.; Yanase, M.; Sekiguchi, Y.; Ogasawara, K. *Tetrahedron Lett.* **1987**, 28, 1783 and references cited therein.
- 4. Kurimura, M.; Takemoto, M.; Achiwa, K. *Chem. Pharm. Bull.* 1991, 39, 2590; Maruyama, Y.; Kurimura, M.; Achiwa. K. *ibid.* 1994, 42, 1709 and references cited therein.
- Murata, M.; Achiwa, K. Chem. Pharm. Bull. 1990, 38, 836; Murata, M.; Ikoma, S.; Achiwa, K. ibid. 1991, 39, 1335.
- Breitogoff, D.; Lawman, K.; Schneider, M. P. J. Chem. Soc., Chem. Commun. 1986, 1523; Murata, M.; Terao, Y.; Achiwa, K.; Nishio, T.; Seto, K. Chem. Pharm. Bull. 1989, 37, 2670; Terao, Y.; Murata, M.; Achiwa, K. Tetrahedron Lett. 1988, 29, 5173.
- Janda, K. D.; Benkovic, S. J.; Lerner, R. A. Science, 1989, 244, 437; Schultz, P. G.; Lerner, R. A. Acc.
  Chem. Res. 1993, 26, 391; Lerner, R. A.; Benkovic, S. J.; Schultz, P. G. Science 1991, 252, 659;

- Ikeda, S.; Weinhouse, M. I.; Janda, K. D.; Lerner, R. A. J. Am. Chem. Soc. 1991, 113, 7763; Kitazume, T.; Lin, J. T.; Takeda, M.; Yamazaki, T. ibid. 1991, 113, 2123; Kitazume, T.; Tsukamoto, T.; Yoshimura, K. J. Chem. Soc. Commun. 1994, 1355; Iwabuchi. Y.; Miyashita, H.; Tanimura, R.; Kinoshita, K.; Kikuchi, M.; Fujii, I. J. Am. Chem. Soc. 1994, 116, 771; Tanaka, F.; Kinoshita, K.; Tanimura, R.; Fujii, I. ibid. 1996, 118, 2332.
- 8. Kohler, G.; Milstein, C. Nature 1975, 256, 495.
- 9. Protein concentration was determined by measurement of the absorbance at 280 nm.
- 10. The enantiomerically pure (*R*)-2 was prepared from (*R*)-3-t-butyldimethylsilyloxy-1-(3-nitrobenzyloxy)glycerol in 2 steps, i) Ac<sub>2</sub>O, pyridine, CH<sub>2</sub>Cl<sub>2</sub>, 93% yield, ii) CF<sub>3</sub>CO<sub>2</sub>H-H<sub>2</sub>O (1:1), 72% yield. (*R*)-2: <sup>1</sup>H NMR(270MHz) δ: 2.10 (s, 3H), 3.57-3.66 (m, 2H), 4.08-4.14 (m, 1H), 4.18-4.22 (m, 2H), 4.68 (s, 2H), 7.50, 8.22 (d, *J*=8.6Hz, each 2H). [α] <sub>D</sub> + 1.0 ° (c=0.28, CHCl<sub>3</sub>).
- 11. Assay conditions:  $100-750 \,\mu\text{M}$  (*R,S*)-1,  $1.3 \,\mu\text{M}$  Ab 2D10 in  $0.2 \,\text{M}$  Tris, pH 8.0,  $30^{\circ}\text{C}$ . Product formation was followed by RP-HPLC ( YMC ODS A-303, 250x4.6 mm, 254 nm, 0.6 mL/min,  $H_2\text{O/CH}_3\text{CN}$  (40:60), 0.05% CF $_3\text{CO}_2\text{H}$ ,  $t_R$  ( Internal Reference ) =  $6.6 \,\text{min}$ ,  $t_R$  ((*R,S*)-1) =  $11.9 \,\text{min}$ ,  $t_R$  ((*R*)-2)=  $7.4 \,\text{min}$ . Retention time of (*S*)-3-acetoxy-1-(3-nitrobenzyloxy)glycerol was  $5.6 \,\text{min}$ .
- 12. The first -order kinetic constant of the background reaction (k<sub>uncat</sub>) was 8.5x10<sup>-3</sup> min<sup>-1</sup> (30°C, pH 8.0).
- 13. The enantioselectivity was measured by HPLC analysis using a column packed with DAICEL CHIRALCEL AD (n-Hexane-IPA=10:1) at 1.0 mL/min. The two enantiomeric products (S)-2 and (R)-2 appeared at 66.9 and 68.6 min, respectively.

(Received in Japan 11 October 1996; accepted 12 December 1996)